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(54) **Hydrofluorocarbon refrigerant compositions soluble in lubricating oil**

(57) Hydrofluorocarbon refrigerants are not soluble in hydrocarbon oil. Retrofitting refrigeration equipment with hydrofluorocarbons generally entails the use of expensive polyol ester (POE) lubricants. The invention provides hydrofluorocarbon refrigerant compositions

that are soluble in hydrocarbon oil thus allowing for a simpler and less expensive retrofit option to the refrigerant user.

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**Description**Field of the Invention

**[0001]** The invention relates generally to compositions for use as refrigerants, aerosol propellants, heat transfer media, gaseous dielectrics, fire-extinguishing agents, foam blowing agents and solvents. More particularly, the invention relates to hydrofluorocarbon refrigerant compositions that are soluble in lubricating oil.

Background of the Invention

**[0002]** Fluorocarbon based fluids have found widespread use in industry for refrigeration, air conditioning and heat pump applications.

**[0003]** Vapor compression cycles are one common form of refrigeration. In its simplest form, the vapor compression cycle involves changing the refrigerant from the liquid to the vapor phase through heat absorption at a low pressure, and then from the vapor to the liquid phase through heat removal at an elevated pressure.

**[0004]** While the primary purpose of refrigeration is to remove energy at low temperature, the primary purpose of a heat pump is to add energy at higher temperature. Heat pumps are considered reverse cycle systems because for heating, the operation of the condenser is inter-changed with that of the refrigeration evaporator.

**[0005]** The art is continually seeking new fluorocarbon based fluids, which offer alternatives for refrigeration and heat pump applications. Currently, of particular interest, are fluorocarbon-based mixtures, which are considered to be environmentally acceptable substitutes for the presently used chlorofluorocarbons. The latter, such as monochlorodifluoromethane (R-22) are suspected of causing environmental problems in connection with the earth's protective ozone layer.

**[0006]** The substitute materials must also possess those properties unique to the chlorofluorocarbons including similar refrigeration characteristics, chemical stability, low toxicity, non-flammability, efficiency in-use and low temperature glides.

**[0007]** By "similar refrigeration characteristics" is meant a vapor pressure that is plus or minus 20 percent of the reference refrigerant at the same temperature.

**[0008]** The characteristic of efficiency in-use is important, for example, in air conditioning and refrigeration where a loss in refrigerant thermodynamic performance or energy efficiency may have secondary environmental impacts through increased fossil fuel usage arising from an increased demand for electrical energy.

**[0009]** Alternative hydrofluorocarbon refrigerants are known. R-507 is an azeotrope-like blend of pentafluoroethane (R-125) and 1,1,1-trifluoroethane (R-143a). R-404A is a non-azeotropic blend of R-125, R-143a and 1,1,1,2-tetrafluoroethane (R-134a). R-407 is a non-azeotropic blend of difluoromethane (R-32), R-125 and R-134a. These alternative refrigerants are available commercially from various sources including Honeywell, DuPont, Atochem and ICI.

**[0010]** R-407C serves as a non-ozone-depleting replacement for R-22 in various air-conditioning applications, as well as in most refrigeration systems including chillers. Since R-407C is a close match to R-22 it also serves as a retrofit fluid in applications where R-22 is generally used.

**[0011]** The widespread commercial use of chlorine-free hydrofluorocarbon refrigerants has been hindered, however, by the lack of commercially adequate lubricants. Mineral oil or alkyl benzenes, which have been used traditionally with R-22, are immiscible with R-407C and must therefore be replaced with expensive polyol ester lubricants. In retrofitting refrigerant systems with hydrofluorocarbon refrigerants it is necessary to drain as much of the oil as possible before adding the replacement lubricant. Often this entails removing the compressor from the system to drain lubricant. A chlorine-free R-22 retrofit fluid that is soluble in lubricating oils would advance the art.

Summary of the Invention

**[0012]** The invention provides refrigerant compositions that are soluble in lubricating oils and are particularly useful as R-22 retrofit fluids.

**[0013]** The compositions of the invention comprise a refrigerant and a solubilizing agent. Preferably the refrigerant is a hydrofluorocarbon refrigerant. Optionally, the compositions of the invention further comprise a lubricating oil selected from the group consisting of mineral or hydrocarbon oil, alkyl benzene oil, white or paraffinic oil and mixtures thereof.

**[0014]** In one embodiment of the invention, there is provided a composition comprising a refrigerant selected from the group consisting of the compounds listed in Table I and mixtures thereof and at least one solubilizing agent selected from the group consisting of the compounds listed in Table II and mixtures thereof.

**[0015]** In another embodiment, there is provided a composition comprising (i) from about 80 to about 99.9 weight percent, preferably from about 90 to about 99.9 weight percent, of a refrigerant selected from the compounds listed in

Table I and mixtures thereof; and (ii) from about 20 to about 0.1 weight percent, preferably from about 10 to about 0.1 weight percent, of a solubilizing agent selected from the compounds listed in Table II and mixtures thereof.

[0016] In yet another embodiment of the invention, there is provided a composition comprising (i) from about 80 to about 99.9 weight percent, preferably from about 90 to about 99.9 weight percent, of a refrigerant selected from Table III; and (ii) from about 20 to about 0.1 weight percent, preferably from about 10 to about 0.1 weight percent, of a solubilizing agent selected from the compounds listed in Table II and mixtures thereof.

[0017] In still another preferred embodiment, there is provided a composition comprising (i) from about 80 to about 99.9 weight percent, preferably from about 90 to about 99.9 weight percent, of a refrigerant selected from Table IV; and (ii) from about 20 to about 0.1 weight percent, preferably from about 10 to about 0.1 weight percent, of a solubilizing agent selected from the compounds listed in Table II and mixtures thereof.

[0018] In a particularly preferred embodiment, there is provided a composition comprising (i) from about 80 to about 99.9 weight percent, preferably from about 90 to about 99.9 weight percent, of a refrigerant selected from the group consisting of R-407C, R-410A, R-404A and R-507A; and (ii) from about 20 to about 0.1 weight percent, preferably from about 10 to about 0.1 weight percent, of a solubilizing agent selected from the group consisting of butane, isobutane, pentane, dimethyl ether, and mixtures thereof.

[0019] When a lubricating oil is present, it is present in an amount of from about 1 to about 60 weight percent, preferably from about 10 to about 50 weight percent, based on the total composition.

[0020] In a process embodiment, there is provided a method for producing refrigeration which comprises condensing a composition of the invention and thereafter evaporating the composition in the vicinity of a body to be cooled.

[0021] In another process embodiment, there is provided a method for producing heating which comprises condensing a composition of the invention in the vicinity of a body to be heated and thereafter evaporating the composition.

#### Detailed Description of the Invention and Preferred Embodiments

[0022] The refrigerant may comprise any one of the compounds listed in Table I or mixtures thereof. Hydrofluorocarbon refrigerants are preferred. The term hydrofluorocarbon refers to compounds composed solely of carbon, hydrogen and fluorine atoms. Of the hydrofluorocarbon refrigerants listed in Table I, R-32, R-143a, R-125, R-134a and mixtures thereof are preferred.

[0023] Refrigerant mixtures are preferred. Representative refrigerant mixtures are listed in Table III. Hydrofluorocarbon refrigerant mixtures are preferred. Of the hydrofluorocarbon refrigerant mixtures listed in Table III, R-404A, R-407C, R-410A and R-507A are preferred. Hydrofluorocarbon refrigerant mixtures selected from Table IV are also preferred.

[0024] As used herein, the term solubilizing agent refers to a compound that increases the solubility of a hydrofluorocarbon refrigerant and a lubricating oil in one another. It will be understood that the solubilizing agent may comprise any one of the compounds listed in Table II or mixtures thereof. Butane, isobutane, pentane and dimethyl ether are preferred. The amount of solubilizing agent is an amount effective to dissolve a sufficient amount of refrigerant in the lubricating oil such that the diluted oil can be transported back to the refrigeration compressor. Typically, the solubilizing agent is present in an amount of from about 0.1 to about 20 weight percent, preferably from about 0.1 to about 10 weight percent, more preferably from about 0.5 to about 5 weight percent, most preferably from about 1 to about 2.5 weight %, based on the total composition.

[0025] As used herein, the term lubricating oil refers to mineral or hydrocarbon oil; alkyl benzene oil; white or paraffinic oil; and mixtures thereof. Suitable lubricating oils are commercially available from various sources (e.g., Capella brand names from Texaco and Suniso brand names from Sun Oil). The chemical compositions and uses of these oils are discussed in detail in the book "Fluorocarbon Refrigerants Handbook" by Ralph C. Downing, Prentice Hall, 1998, pp. 206-270. The amount of lubricating oil is an amount effective to provide acceptable lubrication to the compressor parts for its longevity. An effective amount of lubricating oil is the amount recommended by the equipment manufacturer. Typically, the lubricating oil is present in an amount of from about 1 to about 60 weight percent, more preferably from about 0.5 to about 5 weight percent, most preferably from about 1 to about 2.5 weight %, preferably from about 10 to about 50 based on the total composition.

Table I:

| Refrigerants          |                        |                    |
|-----------------------|------------------------|--------------------|
| Refrigerant Number    | Chemical Name          | Chemical Formula   |
| <i>Methane Series</i> |                        |                    |
| 11                    | Trichlorofluoromethane | CCl <sub>3</sub> F |

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Table I: (continued)

| Refrigerants                    |   |  |
|---------------------------------|---|--|
| Refrigerant Number              | Chemical Name                             | Chemical Formula                               |
| <i>Methane Series</i>           |   |  |
| 12                              | Dichlorodifluoromethane                   | $\text{CCl}_2\text{F}_2$                       |
| 13                              | Chlorotrifluoromethane                    | $\text{CClF}_3$                                |
| 14                              | Tetrafluoromethane (carbon tetrafluoride) | $\text{CF}_4$                                  |
| 21                              | Dichlorofluoromethane                     | $\text{CHCl}_2\text{F}$                        |
| 22                              | Chlorodifluoromethane                     | $\text{CHClF}_2$                               |
| 23                              | Trifluoromethane                          | $\text{CHF}_3$                                 |
| 30                              | Dichloromethane (methylene chloride)      | $\text{CH}_2\text{Cl}_2$                       |
| 31                              | Chlorofluoromethane                       | $\text{CH}_2\text{ClF}$                        |
| 32                              | Difluoromethane (methylene fluoride)      | $\text{CH}_2\text{F}_2$                        |
| 40                              | Chloromethane (methylene fluoride)        | $\text{CH}_2\text{F}_2$                        |
| 41                              | Fluoromethane (methyl fluoride)           | $\text{CH}_3\text{F}$                          |
| <i>Ethane Series</i>            |   |  |
| 113                             | 1,1,2-trichloro-1,2,2-trifluoroethane     | $\text{CCl}_2\text{FCClF}_2$                   |
| 114                             | 1,2-dichloro-1,1,2,2-tetrafluoroethane    | $\text{CClF}_2\text{CClF}_2$                   |
| 115                             | Chloropentafluoroethane                   | $\text{CClF}_2\text{CF}_3$                     |
| 116                             | Hexafluoroethane                          | $\text{CF}_3\text{CF}_3$                       |
| 123                             | 2,2-dichloro-1,1,1-trifluoroethane        | $\text{CHCl}_2\text{CF}_3$                     |
| 124                             | 1-chloro-1,1,1,2-tetrafluoroethane        | $\text{CHClF}_2\text{CF}_3$                    |
| 125                             | Pentafluoroethane                         | $\text{CHF}_2\text{CF}_3$                      |
| 134a                            | 1,1,1,2-tetrafluoroethane                 | $\text{CH}_2\text{FCF}_3$                      |
| 141b                            | 1,1-dichloro-1-fluoroethane               | $\text{CH}_3\text{CCl}_2\text{F}$              |
| 142b                            | 1-chloro-1,1-difluoroethane               | $\text{CH}_3\text{CClF}_2$                     |
| 143a                            | 1,1,1-trifluoroethane                     | $\text{CH}_3\text{CF}_3$                       |
| 152a                            | 1,1-difluoroethane                        | $\text{CH}_3\text{CHF}_2$                      |
| <i>Propane Series</i>           |   |  |
| 218                             | Octafluoropropane                         | $\text{CF}_3\text{CF}_2\text{CF}_3$            |
| 227ea                           | 1,1,1,2,3,3,3-heptafluoropropane          | $\text{CF}_3\text{CHF}_2\text{CF}_3$           |
| 236fa                           | 1,1,1,3,3,3-hexafluoropropane             | $\text{CF}_3\text{CH}_2\text{CF}_3$            |
| 245fa                           | 1,1,1-3,3,3-pentafluoropropane            | $\text{CF}_3\text{CH}_2\text{CHF}_2$           |
| <i>Butane Series</i>            |   |  |
| 365                             | 1,1,1,3,3-pentafluorobutane               | $\text{CF}_3\text{CH}_2\text{CF}_2\text{CH}_3$ |
| <i>Cyclic Organic Compounds</i> |   |  |
| C318                            | Octafluorocyclobutane                     | $-(\text{CF}_2)_4-$                            |

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Table II:

| Solubilizing Agents                    |                                      |  |
|--|--------------------------------------|--|
| Refrigerant Number                     | Chemical Name                        | Chemical Formula   |
| <i>Miscellaneous Organic Compounds</i> |                                      |  |
| <i>Hydrocarbons</i>                    |                                      |  |
| 30                                     | Dichloromethane (methylene chloride) | $\text{CH}_2\text{Cl}_2$   |
| 40                                     | Chloromethane (methyl chloride)      | $\text{CH}_3\text{Cl}$   |
| 50                                     | Methane                              | $\text{CH}_4$  |
| 170                                    | Ethane                               | $\text{CH}_3\text{CH}_3$   |
| 290                                    | Propane                              | $\text{CH}_3\text{CH}_2\text{CH}_3$  |
| 600                                    | Butane                               | $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_3$                                 |
| 600a                                   | Isobutane                            | $\text{CH}(\text{CH}_3)_2\text{CH}_3$  |
| ----                                   | Pentane                              | $\text{CH}_3(\text{CH}_2)_3\text{CH}_3$  |
| ----                                   | Isopentane                           | $\text{CH}_3\text{CH}(\text{CH}_3)\text{CH}_2\text{CH}_3$                      |
| ----                                   | Neopentane                           | $\text{CH}_3\text{C}(\text{CH}_3)_2\text{CH}_3$                                |
| ----                                   | Cyclopentane                         | $-(\text{CH}_2)_5-$  |
| <i>Fluorocarbons</i>                   |                                      |  |
| 1311                                   | Iodotrifluoromethane                 | $\text{CF}_3\text{I}$  |
| ----                                   | Pentafluorodimethyl ether            | $\text{CF}_2\text{OCF}_2\text{H}$  |
| 152a                                   | 1,1-difluoroethane                   | $\text{CH}_3\text{CHF}_2$  |
| 161                                    | Fluoroethane                         | $\text{CH}_3\text{CH}_2\text{F}$   |
| 218                                    | Hexafluoroethane                     | $\text{CF}_3\text{CF}_3$   |
| <i>Oxygen compounds</i>                |                                      |  |
| ----                                   | Dimethyl ether                       | $\text{CH}_3\text{OCH}_3$  |
| 610                                    | Ethyl ether                          | $\text{CH}_3\text{CH}_2\text{OCH}_2\text{CH}_3$                                |
| ----                                   | Poly butylene glycols                | $\text{H}-(\text{O}-\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2)_n-\text{OH}$ |
| 611                                    | Methyl formate                       | $\text{HCOOCH}_3$  |
| <i>Inorganic Compounds</i>             |                                      |  |
| 744                                    | Carbon dioxide                       | $\text{CO}_2$  |
| 764                                    | Sulfur hexafluoride                  | $\text{SF}_6$  |
| <i>Unsaturated Organic Compounds</i>   |                                      |  |
| 1150                                   | Ethene (ethylene)                    | $\text{CH}_2=\text{CH}_2$  |
| 1270                                   | Propene (propylene)                  | $\text{CH}_3\text{CH}=\text{CH}_2$   |

Table III:

| Refrigerant Mixtures |                          |
|----------------------|--------------------------|
| Refrigerant Number   | Composition (Wt.%)       |
| <i>Zeotropes</i>     |                          |
| 400                  | R-12/114 (40/60)         |
| 401A                 | R-22/152a/124 (53/13/34) |

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Table III: (continued)

| Refrigerant Mixtures |                                     |
|----------------------|-------------------------------------|
| Refrigerant Number   | Composition (Wt.%)                  |
| <i>Zeotropes</i>     |                                     |
| 401B                 | R-22/152a/124 (61/11/28)            |
| 401C                 | R-22/152a/124 (33/15/52)            |
| 402A                 | R-125/290/22 (60/2/38)              |
| 402B                 | R-125/290/22 (38/2/60)              |
| 403A                 | R-290/22/218 (5/75/20)              |
| 403B                 | R-290/22/218 (5/56/39)              |
| 404A                 | R-125/143a/134a (44/52/4)           |
| 405A                 | R-22/152a/142b/C318 (45/7/5.5/42.5) |
| 406A                 | R-22/600a/142b (55/4/41)            |
| 407A                 | R-32/125/134a (20/40/40)            |
| 407B                 | R-32/125/134a (10/70/20)            |
| 407C                 | R-32/125/134a (23/25/52)            |
| 407D                 | R-32/125/134a (15/15/70)            |
| 407E                 | R-32/125/134a (25/15/60)            |
| 408A                 | R-125/143a/22 (7/46/47)             |
| 409A                 | R-22/124/142b (60/25/15)            |
| 409B                 | R-22/124/142b (65/25/10)            |
| 410A                 | R-32/125 (50/50)                    |
| 410B                 | R-32/125 (45/55)                    |
| 411A                 | R-1270/22/152a) (1.5/87.5/11.0)     |
| 411B                 | R-1270/22/152a (3/94/3)             |
| 412A                 | R-22/218/143b (70/5/25)             |
| 413A                 | R-218/134a/600a (9/88/3)            |
| 414A                 | R-22/124/600a/142b (51/28.5/4/16.5) |
| 414B                 | R-22/124/600a/142b (50/39/1.5/9.5)  |
| <i>Azeotropes</i>    |                                     |
| 500                  | R-12/152a (73.8/26.2)               |
| 501                  | R-22/12 (75.0/25.0)                 |
| 502                  | R-22/115 (48.8/51.2)                |
| 503                  | R-23/13 (40.1/59.9)                 |
| 504                  | R-32/115 (48.2/51.8)                |
| 505                  | R-12/31 (78.0/22.0)                 |
| 506                  | R-31/114 (55.1/44.9)                |
| 507A                 | R-125/143a (50/50)                  |
| 508A                 | R-23/116 (39/61)                    |
| 508B                 | R-23/116 (46/54)                    |

Table III: (continued)

| Refrigerant Mixtures |                    |
|----------------------|--------------------|
| Refrigerant Number   | Composition (Wt.%) |
| <i>Azeotropes</i>    |                    |
| 509A                 | R-22/218 (44/56)   |

Table IV:

| Hydrofluorocarbon Refrigerant Mixtures |           |            |             |             |
|--|-----------|------------|-------------|-------------|
| Mixture #                              | R-32 wt.% | R-125 wt.% | R-143a wt.% | R-134a wt.% |
| 1                                      | 10 to 80  | 90 to 20   | -           | -           |
| 2                                      | 5 to 45   | 5 to 45    | -           | 90 to 10    |
| 3                                      | -         | 30 to 90   | 70 to 10    | -           |
| 4                                      | -         | 60 to 40   | 39 to 20    | 2 to 40     |

**[0026]** The components of the composition of the invention are known materials that are commercially available or may be prepared by known methods. Preferably, the components are of sufficiently high purity so as to avoid the introduction of adverse influences on the properties of the system.

**[0027]** The compositions of the invention may also contain additives such as oxidation resistance and thermal stability enhancers, corrosion inhibitors, metal deactivators, lubricity additives, viscosity index enhancers, pour and/or floc point depressants, detergents, dispersants, antifoaming agents, anti-wear agents, and extreme pressure resistant additives. Many additives are multifunctional. For example, certain additives may impart both anti-wear and extreme pressure resistance properties, or function both as a metal deactivator and a corrosion inhibitor. Cumulatively, all additives preferably do not exceed 8% by weight, or more preferably do not exceed 5% by weight, of the total composition.

**[0028]** An effective amount of the foregoing additive types is generally in the range from 0.01 to 5% for the antioxidant component, 0.01 to 5% for the corrosion inhibitor component, from 0.001 to 0.5% for the metal deactivator component, from 0.5 to 5% for the lubricity additives, from 0.01 to 2% for each of the viscosity index enhancers and pour and/or floc point depressants, from 0.1 to 5% for each of the detergents and dispersants, from 0.001 to 0.1% for anti-foam agents, and from 0.1-2% for each of the anti-wear and extreme pressure resistance components. All these percentages are by weight and are based on the total composition. It is to be understood that more or less than the stated amounts of additives may be more suitable to particular circumstances, and that a single molecular type or a mixture of types may be used for each type of additive component. Also, the examples listed below are intended to be merely illustrative and not limiting.

**[0029]** Examples of suitable oxidation resistance and thermal stability enhancers are diphenyl-, dinaphthyl-, and phenylnaphthyl-amines, in which the phenyl and naphthyl groups can be substituted, e.g., N,N'-diphenyl phenylenediamine, p-octyldiphenylamine, p,p'-dioctyldiphenylamine, N-phenyl-1-naphthyl amine, N-phenyl-2-naphthyl amine, N-(p-dodecyl)phenyl-2-naphthyl amine, di-1-naphthylamine, and di-2-naphthylamine; phenothiazines such as N-alkyl-phenothiazines; imino(bisbenzyl); and hindered phenols such as 6-(t-butyl) phenol, 2,6-di-(t-butyl) phenol, 4-methyl-2,6-di-(t-butyl) phenol, 4,4'-methylenebis(2,6-di-(t-butyl) phenol), and the like.

**[0030]** Examples of suitable cuprous metal deactivators are imidazole, benzimidazole, 2-mercaptobenzthiazole, 2,5-dimercaptothiadiazole, salicylidine-propylenediamine, pyrazole, benzotriazole, tolutriazole, 2-methylbenzamidazole, 3,5-imethyl pyrazole, and methylene bis-benzotriazole. Benzotriazole derivatives are preferred. Other examples of more general metal deactivators and/or corrosion inhibitors include organic acids and their esters, metal salts, and anhydrides, e.g., N-oleyl-sarcosine, sorbitan mono-oleate, lead naphthenate, dodecenylnsuccinic acid and its partial esters and amides, and 4-nonylphenoxy acetic acid; primary, secondary, and tertiary aliphatic and cycloaliphatic amines and amine salts of organic and inorganic acids, e.g., oil-soluble alkylammonium carboxylates; heterocyclic nitrogen containing compounds, e.g., thiadiazoles, substituted imidazolines, and oxazolines; quinolines, quinones, and anthraquinones; propyl gallate; barium dinonyl naphthalene sulfonate; ester and amide derivatives of alkenyl succinic anhydrides or acids, dithiocarbamates, dithiophosphates; amine salts of alkyl acid phosphates and their derivatives.

**[0031]** Examples of suitable lubricity additives include long chain derivatives of fatty acids and natural oils, such as esters, amines, amides, imidazolines, and borates.

**[0032]** Examples of suitable viscosity index enhancers include polymethacrylates, copolymers of vinyl pyrrolidone and methacrylates, polybutenes, and styrene-acrylate copolymers.

**[0033]** Examples of suitable pour point and/or floc point depressants include polymethacrylates such as methacrylate-ethylene-vinyl acetate terpolymers; alkylated naphthalene derivatives; and products of Friedel-Crafts catalyzed condensation of urea with naphthalene or phenols.

**[0034]** Examples of suitable detergents and/or dispersants include polybutenylsuccinic acid amides; polybutenyl phosphonic acid derivatives; long chain alkyl substituted aromatic sulfonic acids and their salts; and metal salts of alkyl sulfides, of alkyl phenols, and of condensation products of alkyl phenols and aldehydes.

**[0035]** Examples of suitable anti-foam agents include silicone polymers and some acrylates.

**[0036]** Examples of suitable anti-wear and extreme pressure resistance agents include sulfurized fatty acids and fatty acid esters, such as sulfurized octyl tallate; sulfurized terpenes; sulfurized olefins; organopolysulfides; organo phosphorus derivatives including amine phosphates, alkyl acid phosphates, dialkyl phosphates, aminedithiophosphates, trialkyl and triaryl phosphorothionates, trialkyl and triaryl phosphines, and dialkylphosphites, e.g., amine salts of phosphoric acid monoethyl ester, amine salts of dinonylnaphthalene sulfonate, triphenyl phosphate, trinaphthyl phosphate, diphenyl cresyl and dicresyl phenyl phosphates, naphthyl diphenyl phosphate, triphenylphosphorothionate; dithiocarbamates, such as an antimony dialkyl dithiocarbamate; chlorinated and/or fluorinated hydrocarbons, and xanthates.

#### Examples

##### Example 1

**[0037]** Critical flammability ratio (CFR) of mixtures determined using Underwriter Laboratories Refrigerant Flammability test method 2182. CFR allows determination of what level of flammable material can be incorporated, without the mixture itself becoming flammable. Butane and dimethyl ether (DME) were added to R-407C (23 wt.% R-32; 25 wt.% R-125 and 52 wt.% R-134a) as the flammable additive.

| Temp. (°C) | Additive | Wt. % |
|------------|----------|-------|
| 25         | Butane   | 3.4   |
| 60         | Butane   | 3.1   |
| 100        | Butane   | 2.5   |
| 25         | DME      | 3.3   |
| 60         | DME      | 2.7   |
| 100        | DME      | 2.1   |

**[0038]** The table indicates that certain 32/125/134a/butane (or DME) compositions are nonflammable per the widely used refrigerant flammability method.

##### Example 2

**[0039]** Actual testing in a refrigeration machine of a composition of the invention (test mixture: 22.5 wt.% R-32, 24.5 wt.% R-125, 51 wt.% R-134a, 2 wt.% butane) was performed under typical air conditioning conditions and using mineral oil supplied by the compressor manufacturer (Copeland blended white oil Catalog No. 999-5170-31).

**[0040]** Testing was performed in a 2 ton air conditioner system setup similar to the unit reported in Report DOE/CE/23810-71 "Study of Lubricant Circulation in HVAC Systems," March 1995-April 1996 (author Frank R. Biancardi et. al.; prepared for Air Conditioning and Refrigeration Technology Institute Under ARTI/MCLR Project No. 665-53100) except that instead of equal size risers, three different size risers (3/4", 7/8" and 11/8") were used to allow a greater variety of velocities. Also the ability to pump oil from the compressor sump into the compressor discharge line was added. Using a hand pump, 90cc of oil from the sump was injected into the compressor discharge line. By observing the oil level in the compressor sump versus time, the rate and time required for oil return was measured.

| Composition  | Capacity (tons) | COP | Oil return rate (cc/min) | Completion Time (min) |
|--------------|-----------------|-----|--------------------------|-----------------------|
| R-407C       | 1.70            | 2.9 | 2.4                      | 65                    |
| Test Mixture | 1.88            | 3.3 | 2.7                      | 46                    |



**[0041]** Oil return is important for compressor reliability purposes. The example demonstrates that both capacity and efficiency of the system are enhanced over R-407C in the absence of the solubilizing agent and the oil return is improved as well (in mineral oil systems).

#### Examples 3-7

**[0042]** The performance of mixtures of HFC refrigerant R-407C is obtained in the presence and absence of solubilizing agents in a thermodynamic refrigeration cycle operating in typical air conditioning application (100°F condensing temperature and 40°F evaporating temperature). The vapor pressure of each mixture is measured at 25°C.

| Ex. No. | 32/125/134a<br>Refrigerant<br>Mixture (wt.%) | Solubilizing Agent | Wt.% | Refrigeration<br>Capacity relative<br>to R-407C | Vapor Pressure<br>relative to R-407C<br>(25°C) | Oil Return |
|---------|--|--------------------|------|---|--|------------|
| 3       | 23/25/52 (R-407C)                            | --                 | 0    | 1   | 1  | No         |
| 4       | 21.9/23.8/49.3                               | Propane            | 5    | 1.05  | 1.05   | Yes        |
| 5       | 21.9/23.8/49.3                               | Butane             | 5    | 1.00  | 0.99   | Yes        |
| 6       | 22.6/24.5/51.0                               | Butane             | 2    | 1.00  | 1.00   | Yes        |
| 7       | 20.9/22.7/47.4                               | Butane             | 10   | 1.00  | 0.98   | Yes        |

**[0043]** Example 3 demonstrates that the HFC itself is unsuitable for use with hydrocarbon oil in that without oil return the compressor may be damaged.

**[0044]** Example 4 demonstrates that although there is oil return, the addition of propane results in an undesirable increase in the pressure of the system.

**[0045]** Examples 5-7 demonstrate that the use of butane results in oil return without an undesirable pressure increase. This is particularly important for retrofitting applications.

#### Claims

1. A composition comprising:

- (i) a hydrofluorocarbon refrigerant; and
- (ii) a solubilizing agent selected from the group consisting of butane, isobutane, pentane, dimethyl ether and mixtures thereof.

2. The composition of claim 1 wherein the hydrofluorocarbon refrigerant comprises a compound selected from the group consisting of R-32, R-125, R-134a, R-143a and mixtures thereof.

3. The composition of claim 1 wherein the hydrofluorocarbon refrigerant is a mixture of R-32, R-125 and R-134a.

4. The composition of claim 1 wherein the hydrofluorocarbon refrigerant is a mixture of R-32 and R-125.

5. The composition of claim 1 wherein the hydrofluorocarbon refrigerant is a mixture of R-143a, R-125 and R-134a.

6. The composition of claim 1 wherein the hydrofluorocarbon refrigerant is a mixture of R-125 and R-143a.

7. The composition of claim 1 wherein the hydrofluorocarbon refrigerant is selected from the group consisting of the compositions listed in Table III.

8. The composition of claim 1 wherein the hydrofluorocarbon refrigerant is selected from the group consisting of R-404A, R-407C, R-410A and R-507A.

9. The composition of claim 1 wherein the hydrofluorocarbon refrigerant is selected from the group consisting of the compositions listed in Table IV.

10. The composition of any one of claims 1-9 wherein the solubilizing agent is butane.

11. The composition of any one of claims 1-9 wherein the solubilizing agent is isobutane.

5 12. The composition of claim 1 which comprises about 80-99.9 weight percent of R407C and about 0.1-10 weight percent of butane.

13. The composition of claim 1 which comprises about 80-99.9 weight percent of R410A and about 0.1-10 weight percent of butane.

10 14. The composition of any one of claims 1-9 wherein the solubilizing agent is pentane.

15. The composition of any one of claims 1-9 wherein the solubilizing agent is dimethyl ether.

15 16. The composition of any preceding claim further comprising a lubricating oil selected from the group consisting of mineral or hydrocarbon oil, alkyl benzene oil, white or paraffinic oil and mixtures thereof.

17. A method for producing refrigeration which comprises condensing a composition as defined in any of claims 1 to 15 and thereafter evaporating the composition in the vicinity of a body to be cooled.

20 18. A method for producing heating which comprises condensing a composition as defined in any of claims 1 to 15 in the vicinity of a body to be heated and thereafter evaporating the composition.

25 19. A method of recharging a refrigeration system of the type containing a chlorine-containing refrigerant and a lubricating oil, comprising the steps of:

(a) removing said chlorine-containing refrigerant from said refrigeration system while leaving substantially all of said lubricating oil in said system; and

30 (b) introducing to said lubricating oil left in said system a composition as defined in any of claims 1 to 15.

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European Patent  
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## EUROPEAN SEARCH REPORT

Application Number  
EP 00 31 0474

## DOCUMENTS CONSIDERED TO BE RELEVANT

| Category  | Citation of document with indication, where appropriate, of relevant passages  | Relevant to claim    | CLASSIFICATION OF THE APPLICATION (Int.Cl.7) |
|---|--|----------------------|--|
| X   | EP 1 018 538 A (NIPPON MITSUBISHI OIL CORP) 12 July 2000 (2000-07-12)<br>* paragraph [0008] - paragraph [0009] *<br>* paragraph [0102] - paragraph [0106];<br>tables 6,7 *                         | 1-13,<br>16-18       | C09K5/04                                     |
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| X   | WO 96 03473 A (MURPHY FREDERICK THOMAS ; CORR STUART (GB); ICI PLC (GB); MORRISON)<br>8 February 1996 (1996-02-08)<br>* page 3, line 5 - page 5, line 23; claims 1-21; examples 1-12; tables 1-8 * | 1-9,14,<br>16-18     |  |
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|   | ---  |                      |  |
|   | -/--   |                      |  |
| The present search report has been drawn up for all claims  |  |                      |  |
| Place of search   | Date of completion of the search   | Examiner             |  |
| MUNICH  | 23 May 2001  | olde Scheper, B      |  |
| CATEGORY OF CITED DOCUMENTS   |  |                      |  |
| X : particularly relevant if taken alone<br>Y : particularly relevant if combined with another document of the same category<br>A : technological background<br>O : non-written disclosure<br>P : intermediate document<br>T : theory or principle underlying the invention<br>E : earlier patent document, but published on, or after the filing date<br>D : document cited in the application<br>L : document cited for other reasons<br>& : member of the same patent family, corresponding document |  |                      |  |

EPO FORM 1503 03.02 (P04C01)



European Patent  
Office

## EUROPEAN SEARCH REPORT

Application Number  
EP 00 31 0474

| DOCUMENTS CONSIDERED TO BE RELEVANT   |   |   |  |
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| Category  | Citation of document with indication, where appropriate, of relevant passages                           | Relevant to claim                               | CLASSIFICATION OF THE APPLICATION (Int.Cl.7) |
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|   |   |   | TECHNICAL FIELDS SEARCHED (Int.Cl.7)         |
|   |   |   |  |
| The present search report has been drawn up for all claims  |   |   |  |
| Place of search<br>MUNICH   |   | Date of completion of the search<br>23 May 2001 | Examiner<br>olde Scheper, B                  |
| <p>CATEGORY OF CITED DOCUMENTS</p> <p>X : particularly relevant if taken alone<br/> Y : particularly relevant if combined with another document of the same category<br/> A : technological background<br/> O : non-written disclosure<br/> P : intermediate document</p> <p>T : theory or principle underlying the invention<br/> E : earlier patent document, but published on, or after the filing date<br/> D : document cited in the application<br/> L : document cited for other reasons</p> <p>&amp; : member of the same patent family, corresponding document</p> |   |   |  |

EPO FORM 1503 03/92 (P44C01)

**ANNEX TO THE EUROPEAN SEARCH REPORT  
ON EUROPEAN PATENT APPLICATION NO.**

EP 00 31 0474

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report.  
The members are as contained in the European Patent Office EDP file on  
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

23-05-2001

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EPO FORM P0459

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82

**DERWENT-ACC-NO:** 2002-464721

**DERWENT-WEEK:** 200250

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**TITLE:** Refrigerant composition, e.g. for refrigeration, air conditioning and heat pump applications, comprises hydrofluorocarbon refrigerant and solubilizing agent

**INVENTOR:** RICHARD R G; SINGH R R ; THOMAS R H P ; WILSON D P

**PATENT-ASSIGNEE:** HONEYWELL INT INC[HONE]

**PRIORITY-DATA:** 2000US-670738 (September 27, 2000)

**PATENT-FAMILY:**

| <b>PUB-NO</b> | <b>PUB-DATE</b> | <b>LANGUAGE</b> |
|---------------|-----------------|-----------------|
| EP 1193305 A1 | April 3, 2002   | EN              |

**DESIGNATED-STATES:** AL AT BE CH CY DE DK ES FI FR GB GR IE IT LI LT LU LV  
MC MK NL PT RO SE SI TR

**APPLICATION-DATA:**

| <b>PUB-NO</b> | <b>APPL-DESCRIPTOR</b> | <b>APPL-NO</b> | <b>APPL-DATE</b>  |
|---------------|------------------------|----------------|-------------------|
| EP 1193305A1  | N/A                    | 2000EP-310474  | November 24, 2000 |

**INT-CL-CURRENT:**

| <b>TYPE</b> | <b>IPC DATE</b>     |
|-------------|---------------------|
| CIPS        | C09K5/04 20060101   |
| CIPS        | C10M171/00 20060101 |

**RELATED-ACC-NO:** 2002-463198 2004-282574

**ABSTRACTED-PUB-NO:** EP 1193305 A1

**BASIC-ABSTRACT:**

**NOVELTY** - Refrigerant composition comprises a hydrofluorocarbon refrigerant; and a solubilizing agent including butane, isobutane, pentane, dimethyl ether, or their mixtures.

**DESCRIPTION - INDEPENDENT CLAIMS** are also included for;

(1) A method for producing refrigeration, which involves condensing the above composition, and evaporating the composition in the vicinity of a body to be cooled;

(2) A method for producing heating, which involves condensing the inventive composition in the vicinity of a body to be heated, and evaporating the composition; and

(3) A method of recharging a refrigeration system of the type containing a chlorine-containing refrigerant and a lubricating oil, which involves removing the chlorine-containing refrigerant from the refrigeration system while leaving all of the lubricating oil in the system, and introducing to the lubricating oil left in the system the inventive composition.

**USE** - The refrigerant composition is used in industry for refrigeration, air conditioning, and heat pump applications. It is used as refrigerants, aerosol propellants, heat transfer media, gaseous dielectrics, fire-extinguishing agents, foam blowing agents, and solvents. It is used as R-22 retrofit fluids.

**ADVANTAGE** - The inventive refrigerant composition is soluble in hydrocarbon oil, thus allowing for a simpler and less expensive retrofit option to the refrigerant user.

## **EQUIVALENT-ABSTRACTS:**

### **ORGANIC CHEMISTRY**

**Preferred Components:** The hydrofluorocarbon refrigerant comprises a compound including R-32, R-125, R-134a, R-143a, or their mixtures. It can be one of the compositions listed in Table III and IV as disclosed. It includes R-404A, R-407C, R-410A, or R-507A. The solubilizing agent is butane or isobutane. It can be pentane or dimethyl ether. The composition further comprises a lubricating oil including mineral or hydrocarbon oil, alkyl benzene oil, white or paraffinic oil, or their mixtures.

**Preferred Composition:** The composition comprises 80-99.9 wt.% R407C or R410A and 0.1-10 wt.% butane.

Actual testing in a refrigeration machine of the inventive composition comprising 22.5 wt.% R-32, 24.5 wt.% R-125, 51 wt.% R-134a, and 2 wt.% butane was performed. Testing was performed in a 2 ton air conditioner system setup similar to the unit reported in Report DOE/CE/23810-71 Study of Lubricant Circulation in HvAC Systems, March 1995-April 1996 (author Frank R. Biancardi et. al., prepared for Air Conditioning and Refrigeration Technology Institute Under ARTI/MCLR Project No. 665-53100 except that instead of equal size risers, three different size risers (3/4, 7/8, and 1 1/8) were used. Using a hand pump, 90

cc of oil from the sump was injected into the compressor discharge line. The rate and time required for oil return was respectively measured to be 2.7 cc/min and 46 minutes for the test mixture, and 2.4 cc/min and 65 minutes for R-407C, demonstrating that both capacity and efficiency of the system were enhanced over R-407C and the oil return was improved.

**TITLE-TERMS:** REFRIGERATE COMPOSITION AIR CONDITION HEAT PUMP APPLY  
COMPRISE SOLUBLE AGENT

**DERWENT-CLASS:** B07 D21 E19 G04 H07 H08 J07 K01 L03 X27

**CPI-CODES:** B04-B01C3; B10-H01; B10-H02B; B10-J02; B11-C09; D08-B10;  
E10-H01E; E10-H04A3; E10-J02D2; E10-J02D3; G04-B01; H07-X;  
H08-D09; J07-A08; K01-A; L03-D01;

**EPI-CODES:** X27-F;

**CHEMICAL-CODES:** Chemical Indexing M1 \*01\* Fragmentation Code M210 M211 M212  
M213 M214 M215 M216 M220 M221 M222 M223 M224 M225 M226  
M231 M232 M233 M320 M416 M423 M430 M610 M620 M782 Q140  
Q337 Q432 Q433 Q434 Q435 Specific Compounds RA00NG  
Registry Numbers 103243

Chemical Indexing M2 \*02\* Fragmentation Code M210 M214 M231  
M320 M416 M430 M610 M620 M782 Q140 Q337 Q432 Q433 Q434  
Q435 Specific Compounds R00804 Registry Numbers 8441

Chemical Indexing M2 \*03\* Fragmentation Code M210 M214 M232  
M320 M416 M430 M610 M620 M782 Q140 Q337 Q432 Q433 Q434  
Q435 Specific Compounds R00355 Registry Numbers 2973

Chemical Indexing M2 \*04\* Fragmentation Code M210 M215 M231  
M320 M416 M430 M610 M620 M782 Q140 Q337 Q432 Q433 Q434  
Q435 Specific Compounds R00879 Registry Numbers 41

Chemical Indexing M2 \*05\* Fragmentation Code H5 H581 H8 M210  
M211 M272 M282 M320 M416 M430 M620 M782 Q140 Q337 Q432  
Q433 Q434 Q435 Specific Compounds R00965 Registry Numbers  
383

Chemical Indexing M2 \*06\* Fragmentation Code H6 H601 H608  
H684 H685 M280 M312 M321 M332 M344 M363 M391 M416 M430  
M620 M782 Q140 Q337 Q432 Q433 Q434 Q435 Specific  
Compounds R16771 Registry Numbers 133559

Chemical Indexing M2 \*07\* Fragmentation Code H6 H601 H681  
H685 M280 M312 M321 M332 M344 M363 M391 M416 M430 M620  
M782 Q140 Q337 Q432 Q433 Q434 Q435 Specific Compounds  
R16596 Registry Numbers 11573



Chemical Indexing M2 \*08\* Fragmentation Code H6 H685 M280  
M312 M321 M331 M343 M363 M391 M416 M430 M620 M782 Q140  
Q337 Q432 Q433 Q434 Q435 Specific Compounds R11342  
Registry Numbers 11569

Chemical Indexing M3 \*02\* Fragmentation Code M210 M214 M231  
M320 M416 M430 M610 M620 M782 Q140 Q337 Q432 Q433 Q434  
Q435 Specific Compounds R00804 Registry Numbers 8441

Chemical Indexing M3 \*03\* Fragmentation Code M210 M214 M232  
M320 M416 M430 M610 M620 M782 Q140 Q337 Q432 Q433 Q434  
Q435 Specific Compounds R00355 Registry Numbers 2973

Chemical Indexing M3 \*04\* Fragmentation Code M210 M215 M231  
M320 M416 M430 M610 M620 M782 Q140 Q337 Q432 Q433 Q434  
Q435 Specific Compounds R00879 Registry Numbers 41

Chemical Indexing M3 \*05\* Fragmentation Code H5 H581 H8 M210  
M211 M272 M282 M320 M416 M430 M620 M782 Q140 Q337 Q432  
Q433 Q434 Q435 Specific Compounds R00965 Registry Numbers  
383

Chemical Indexing M3 \*06\* Fragmentation Code H6 H601 H608  
H684 H685 M280 M312 M321 M332 M344 M363 M391 M416 M430  
M620 M782 Q140 Q337 Q432 Q433 Q434 Q435 Specific  
Compounds R16771 Registry Numbers 133559

Chemical Indexing M3 \*07\* Fragmentation Code H6 H601 H681  
H685 M280 M312 M321 M332 M344 M363 M391 M416 M430 M620  
M782 Q140 Q337 Q432 Q433 Q434 Q435 Specific Compounds  
R16596 Registry Numbers 11573

Chemical Indexing M3 \*08\* Fragmentation Code H6 H685 M280  
M312 M321 M331 M343 M363 M391 M416 M430 M620 M782 Q140  
Q337 Q432 Q433 Q434 Q435 Specific Compounds R11342  
Registry Numbers 11569

Chemical Indexing M6 \*09\* Fragmentation Code Q140 Q254 Q337  
Q432 Q433 Q434 Q435 R524

**UNLINKED-DERWENT-REGISTRY-NUMBERS:** ; 0355U ; 0804U ; 0879U ; 1781U

**SECONDARY-ACC-NO:**

**CPI Secondary Accession Numbers:** 2002-132346